

Ornicorrugatin, a New Siderophore from *Pseudomonas fluorescens* AF76

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From a pyoverdine-negative mutant of *Pseudomonas fluorescens* AF76 a new lipopeptidic siderophore (ornicorrugatin) could be isolated. It is structurally related to the siderophore of *Pseudomonas corrugata* differing in the replacement of one Dab unit by Orn.

Key words: *Pseudomonas fluorescens* AF76, Siderophores, Ornicorrugatin

Introduction

The strain *Pseudomonas fluorescens* AF76 was isolated from the rhizosphere of *Arachis hypogaea* L. in India. A pyoverdine-negative mutant of AF76 showed strong ability to decolourize chrome azurol S (CAS) indicating the production of a secondary siderophore. The structure elucidation of this secondary siderophore, ornicorrugatin (corrugatin where one Dab unit is replaced by Orn), will be reported here.

Materials and Methods

Ornicorrugatin was obtained from the supernatant of a 40-h-old culture of the pyoverdine-negative mutant 1G10 of *Pseudomonas fluorescens* AF76 grown in a casamino acid medium. The supernatant was passed on a C-18 column (3 × 1 cm) and washed twice with distilled water. The siderophore was eluted with H₂O/CH₃CN 4:6. The CAS-positive fraction (Schwyn and Neilands, 1987) was collected and purified by HPLC. Purification was performed on a Gilson system with a 712 HPLC System Controller. A Supelco Discovery[®] BIO Wide Pore column (C-18, 25 × 2.12 cm, 10 μm particle size) was used with a flow rate of 20 ml/min and a gradient going from H₂O/CH₃CN

9:1 containing 0.1% CF₃COOH to H₂O/CH₃CN 2:8 containing 0.1% CF₃COOH in 30 min, followed by 10 min isocratic elution with H₂O/CH₃CN 2:8 containing 0.1% CF₃COOH. From the extract CH₃CN was evaporated *in vacuo* and the sample was lyophilized.

Mass spectral data were obtained with a MAT 900 ST instrument providing an electrostatic/magnetic analyzer (EB) geometry connected to an octapole collision cell and a quadrupole ion trap (QIT), and equipped with an ESI II ion source (Finnigan MAT, Bremen, Germany); spray voltage, 3.4–3.6 kV; capillary temperature, 230 °C. Source conditions were set to minimize fragmentation, resolution ca. 5000 (10% valley). The samples were dissolved in water/methanol/trifluoroacetic acid 50:50:0.1 (v/v). Fragmentation induced by low energy collision activation (CA) was effected in the octapole unit and in the QIT (~2 · 10⁻³ Pa He as bath gas diffusing in the collision octapole). Exact ion mass measurements were performed with an LTQ Orbitrap XL (ThermoFisher, Bremen, Germany) instrument with static nano-ESI (needles with 5 μm inner diameter; Mascom, Bremen, Germany). The resolution (full signal width at half height, FWHH) was 60000 at *m/z* 400 in single stage ESI and 30000 for MS/MS product ion exact mass measurements. The mass accuracy was determined to be < 3 ppm with external calibration.

High resolution ¹H NMR (250 MHz) and ¹³C NMR (62.90 MHz) spectra were recorded in CD₃OD or DMSO-*d*₆ on a Bruker Avance DRX

Abbreviations: Common amino acids, three letter code; Dab, 2,4-diaminobutanoic acid; OHAsp, *threo*-β-hydroxy Asp; OHHis, *threo*-β-hydroxy His; CAS, chrome azurol S; ESI, electrospray ionization; CA, collision activation.

250 spectrometer. Chemical shifts are reported in ppm downfield from TMS.

For chiral amino acid analysis, after hydrolysis (6 M HCl, 110 °C, 24 h), the amino acids were derivatized according to the method described by Demange *et al.* (1988), giving *N*-pentafluoropropionyl (PFP) *O*-trimethylsilyl (TMS) esters. 1 μ l of the toluene solution of each derivatized amino acid was injected in a Hewlett Packard HP6890 gas chromatograph equipped with an Alltech Chirasil-Val column no. 13636 (25 m \times 0.25 mm ID \times 0.16 μ m) and flame ionization detection. Heating program was 4 min at 90 °C, then 4 °C/min to 200 °C.

Results

The octapole CA spectrum (Fig. 1) of corrugatin (**1**, Fig. 2, $n = 2$) (Risse *et al.*, 1998) comprises two parts. Starting from $[M+H]^+$ (m/z 998) losses of up to three molecules of H₂O are observed (m/z 980, 962, 944) as well as of the side chains from the condensation products of β -hydroxy Asp (OHAsp) and β -hydroxy His (OHHis) with Dab, respectively (see Fig 3, loss of 74 Da, m/z 924 and of 96 Da, m/z 902; the latter ion is even more pronounced in the ion trap CA spectrum). These two ions also lose H₂O (m/z 906 and 884, respectively). Loss of both residues results in m/z 828 (subsequent loss of water gives m/z 810 and 792). Loss

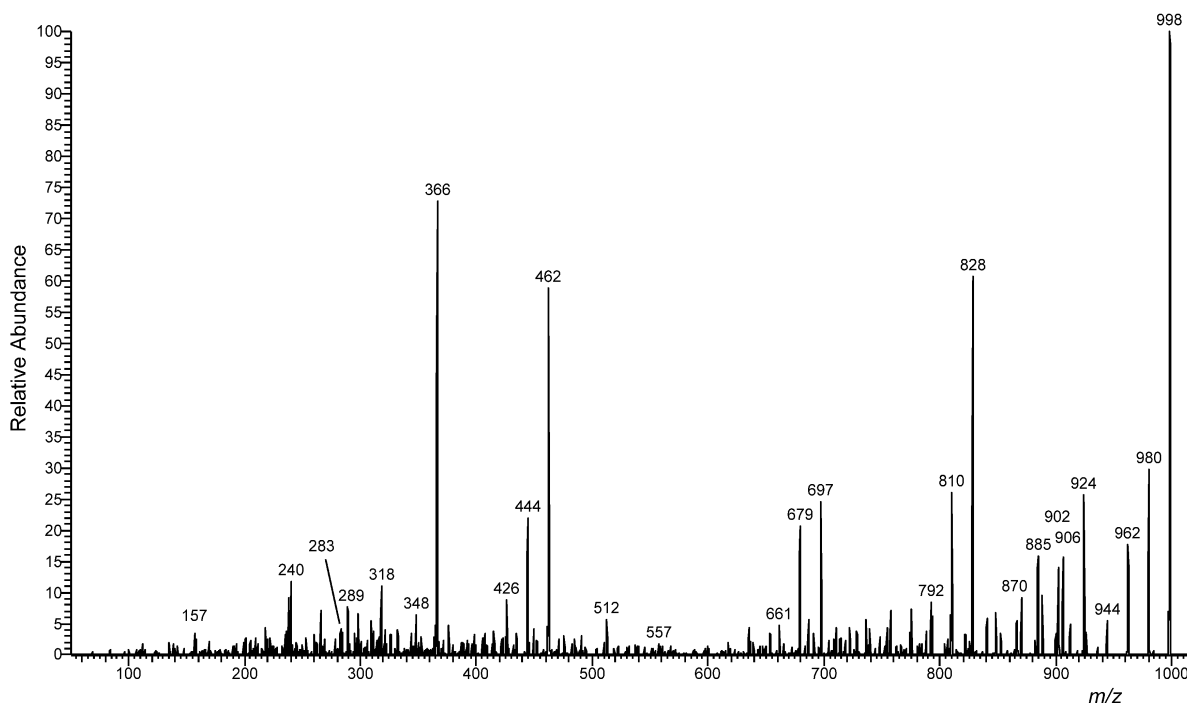


Fig. 1. Octapole CA spectrum of $[M+H]^+$ of corrugatin (**1**).

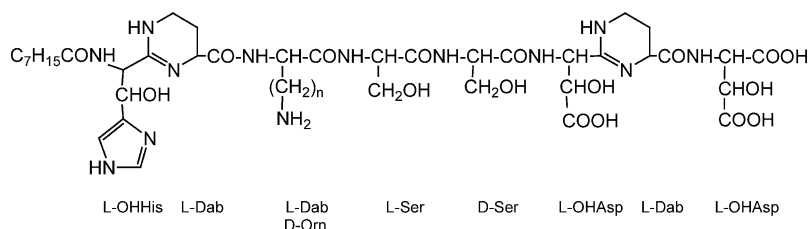


Fig. 2. Corrugatin (**1**, $n = 2$, Dab) and ornicorrugatin (**2**, $n = 3$, Orn).

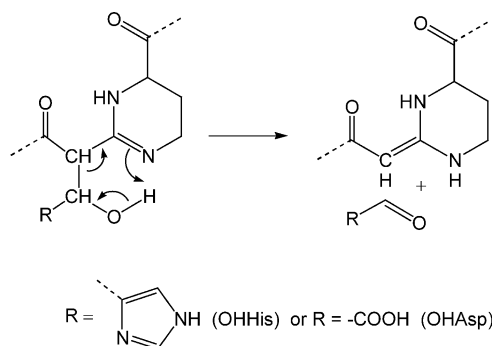


Fig. 3. McLafferty type elimination of amino acid side chains.

of the C-terminal OHAsp with back-transfer of the hydroxy group (Fuchs and Budzikiewicz, 2001) yields the ion m/z 867 (again more pronounced in the ion trap CA spectrum). Loss of (74 + 96) Da from m/z 867 results in m/z 697 ($-H_2O$ gives m/z 679 and 661). Cleavage after Ser₂ with OH back-transfer yields m/z 653 of low abundance (loss of H_2O gives m/z 635).

The most pronounced ions in the lower part of Fig. 1 are m/z 462 (B₃, cleavage after Dab; for a

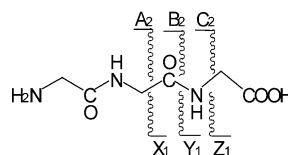


Fig. 4. Designation of peptide fragments. Hyphens (as in Yⁿ) indicate additional H atoms.

designation of peptide fragments see Fig. 4, Roepstorff and Fohlman, 1984) which loses twice H_2O (m/z 444 and 426) and 96 Da (m/z 366, OHHis residue, Fig. 3). Of importance for the subsequent discussion are the ions m/z 318 and 240 formed by the loss of $CH_3(CH_2)_5CH=CO$ (126 Da, the typical keten elimination from amides) from m/z 444 and 366, respectively (Budzikiewicz *et al.*, 1967).

In the octapole CA spectrum of $[M+2H]^{2+}$ (m/z 506.5) Y₁ⁿ (m/z 150) and Y₅ⁿ (m/z 537) can be seen. Important are the ions m/z 444 and 366 (cleavage products of B₃, see above) from which the loss of 126 Da (m/z 318 and 240) is much more pronounced than in the singly charged spectrum). A further ion occurs at m/z 283 (C₂ⁿ - 96 Da, more pronounced than in Fig. 1) which also loses 126

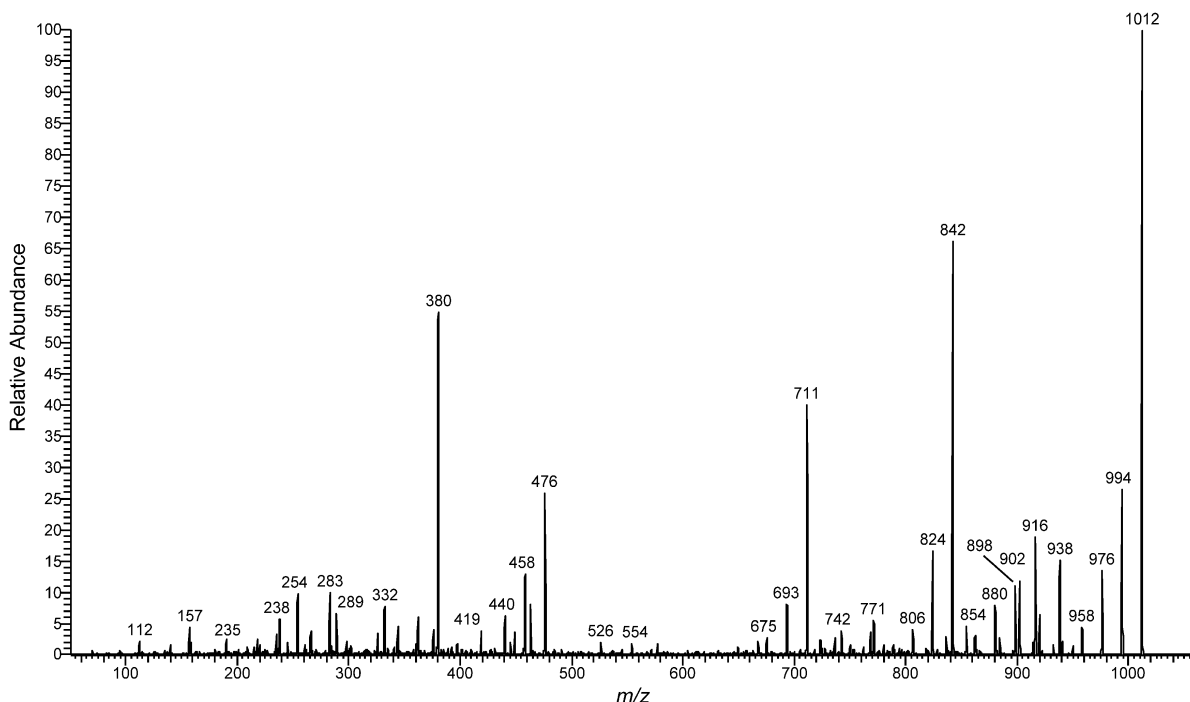


Fig. 5. Octapole CA spectrum of $[M+H]^+$ of ornicorrugatin (2).

Da (m/z 157). $B_2 - 96$ Da (m/z 266) is of lower abundance. In the corresponding mass spectra of corrugatin labeled with ^{15}N all ions show the expected shift values.

The octapole CA spectrum (Fig. 5) of ornicorrugatin (**2**, Fig. 2, $n = 3$) shows striking similarities with that of corrugatin (Fig. 1), but differs in the following way. The molecular mass is 14 Da higher. This suggests either one additional CH_2 group or a replacement of a CH_2 group by CO. The entire upper group of ions is shifted by 14 Da. It follows that the additional 14 Da can not be located in the last three C-terminal amino acids. Since m/z 462 and its degradation products characteristic for the lower group of ions are also shifted in mass the additional unit must be located in B_3 . The presence of m/z 283 and 157 (fragments of C_2'') with identical masses in both spectra demonstrates the absence of the additional 14 Da in these ions. The additional group must therefore be located in the third amino acid. Replacement of Dab by Orn is the most reasonable explanation (Fig. 2, $n = 3$) and is substantiated by the data presented below. The elemental composition of the major ions could be confirmed by exact mass measurements (Table I).

Further evidence is offered by the loss of 126 Da in both spectra from fragments containing the N-terminus indicating the presence of the octanoic acid amide structure, and by the presence of the Y_1' and Y_5'' ions with identical masses for both compounds which indicates that the additional 14 Da can not be located in the five C-terminal amino acids.

The ^1H and ^{13}C NMR spectra of ornicorrugatin (**2**) correspond to those of corrugatin (**1**) (Risse *et*

al., 1998) with the exception that one Dab sequence is replaced by the Orn sequence (α : 4.37/53.0, β : 1.97/22.0, γ : 1.77/24.8, δ : 2.97/40.0 ppm as established by H,H-COSY and HMQC).

Comparison of the free amino acids obtained by acid hydrolysis of authentic (Risse *et al.*, 1998) corrugatin and of ornicorrugatin by gas chromatography on a chiral column after derivatization established an identical composition with the exception that ornicorrugatin contained an additional D-Orn unit. The placement of L- and D-Ser in the peptide chain was established for corrugatin only.

Discussion

So far only twice lipopeptidic siderophores have been described as obtained from fluorescent *Pseudomonas* spp. (Budzikiewicz, 2004), the ferrocins, cyclo-depsidekapeptides with slight variations in the peptide chain (Tsubotani *et al.*, 1993) from *Pseudomonas fluorescens* YK-310, and corrugatin (Risse *et al.*, 1998) from *Pseudomonas corrugata*. The taxonomical placement of this species has been controversial, but currently it is placed in close vicinity to the fluorescent *Pseudomonas* spp. (Sutra *et al.*, 1997). The isolation of the structurally closely related ornicorrugatin from a *Pseudomonas fluorescens* strain favours this placement.

The main siderophores of the fluorescent *Pseudomonas* spp. are the pyoverdins, chromopeptides comprising a dihydroxyquinoline chromophore and a peptide chain consisting of six to twelve amino acids, partially modified (Budzikiewicz, 2004). In addition, secondary siderophores with lower ability to bind Fe^{3+} are produced, especially by pyoverdin-negative strains. A siderophore encountered with several *Pseudomonas* species is pyochelin derived from salicylic acid and two molecules of cysteine (Cobessi *et al.*, 2005; Schlegel *et al.*, 2006; see also Budzikiewicz, 2004). Ornicorrugatin is another secondary siderophore which had been overlooked so far.

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Table I. Exact mass data of selected ions from Fig. 3.

Mass found	Mass calcd.	Composition
1012.470	1012.470	$\text{C}_{41}\text{H}_{66}\text{N}_{13}\text{O}_{17}$
842.435	842.437	$\text{C}_{35}\text{H}_{60}\text{N}_{11}\text{O}_{13}$
476.296	496.299	$\text{C}_{23}\text{H}_{38}\text{N}_7\text{O}_4$
458.286	458.288	$\text{C}_{23}\text{H}_{36}\text{N}_7\text{O}_3$
380.264	380.266	$\text{C}_{19}\text{H}_{34}\text{N}_5\text{O}_3$

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